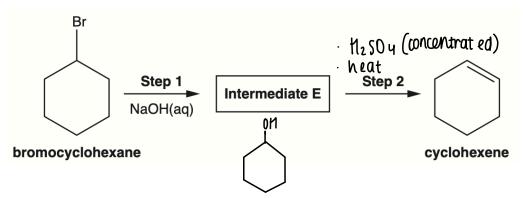


AS level Chemistry A

H032/02 Depth in chemistry

Question Set 6

- Organic compounds can be prepared in the laboratory using synthetic routes with two or more stages.
 - (a) A student devises a two-stage synthesis of cyclohexene from bromocyclohexane.



- (i) Suggest the structure of **intermediate E** and the reagent(s) and conditions for **step 2**.
- (ii) The student carries out this synthesis and obtains 1.23 g of pure cyclohexene from 5.50 g of bromocyclohexane.

[2]

[3]

Calculate the percentage yield of cyclohexene.

Give your final answer to an **appropriate** number of significant figures.

Moles bromocyclonexane:
$$\frac{5.50}{162.9}$$
 = 0.033763

theoretical mass of (yclon exerce = 0.033763×82 = 2.77g/· yield = $\frac{1.23}{2.77} \times 100 = 44.4 / ...$

(b) Cyclohexene is reacted with bromine to prepare the organic compound **F**.

Give the structure of compound **F** and outline the mechanism for this reaction.

Include curly arrows, charges and relevant dipoles.

Total Marks for Question Set 6: 9



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